

Six Non-equivalent Nitrogen Atoms in Octahedral mer -[Fe(fbpy)₃]²⁺ (fbpy = 4-fluoro-2,2'-bipyridine)

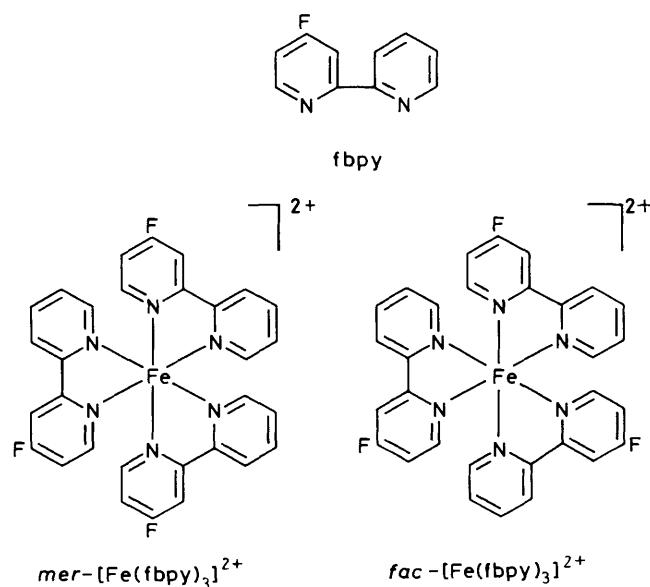
A. F. Janzen,* T. O. Nguyen, Fanqi Qu, and K. Marat

Department of Chemistry, University of Manitoba, Winnipeg, Manitoba, Canada R3T 2N2

The ¹⁵N n.m.r. spectra of fbpy (4-fluoro-2,2'-bipyridine), [Fe(bpy)₃]²⁺, cis -[Co(bpy)₂(H₂O)₂]³⁺, and a 3:1 mixture of mer - and fac -[Fe(fbpy)₃]²⁺, are reported; in the case of mer -[Fe(fbpy)₃]²⁺, the lack of symmetry makes all six nitrogen atoms non-equivalent and this non-equivalence is confirmed by ¹⁵N n.m.r. spectroscopy.

Fay and Piper¹ found that a mixture of mer - and fac -isomers of a metal trifluoroacetylacetonate, M(tfac)₃, gave four CF₃ signals in the ¹⁹F n.m.r. spectrum, three from the mer - and one from the fac -isomer. As part of our studies of fluorine exchange in various trigonal bipyramidal and octahedral compounds,² we synthesized 4-fluoro-2,2'-bipyridine (fbpy) and prepared a series of octahedral complexes containing the fbpy or bpy (2,2'-bipyridine) ligand. A statistical 3:1 mixture of mer - and fac -[Fe(fbpy)₃]²⁺ showed³ four signals of equal intensity in the ¹⁹F n.m.r. spectrum at -97.69, -97.72, -97.78, and -97.82 p.p.m., three assigned to mer -[Fe(fbpy)₃]²⁺ and one to fac -[Fe(fbpy)₃]²⁺, but these isomers have not, so far, been separated. mer -[Fe(fbpy)₃]²⁺ is of particular interest because its lack of symmetry makes all six nitrogen atoms in the octahedral inner co-ordination sphere non-equivalent; we therefore examined the ¹⁵N n.m.r. spectrum to see if this non-equivalence could be observed.

The ¹⁵N n.m.r. spectra of several bpy and fbpy complexes are shown in Figure 1. As expected, [Fe(bpy)₃]²⁺ gives a single peak [Figure 1(a)], and cis -[Co(bpy)₂(H₂O)₂]³⁺ gives two singlet peaks [Figure 1(b)], while the non-equivalent nitrogen atoms of fbpy [Figure 1(c)] give rise to a singlet ('pyridyl') and



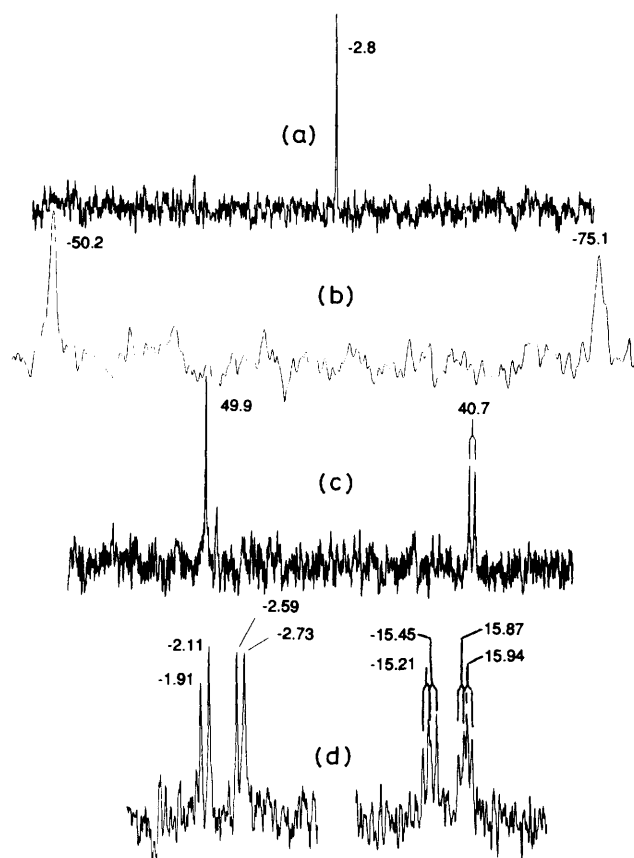


Figure 1. Natural abundance, proton-decoupled, ^{15}N n.m.r. spectra recorded with a Bruker AM300 spectrometer at 30.4 MHz: (a) $[\text{Fe}(\text{bpy})_3]\text{Cl}_2$ in D_2O ; (b) $\text{cis-}[\text{Co}(\text{bpy})_2(\text{H}_2\text{O})_2](\text{PF}_6)_3$ in D_2O ; (c) fbpy in CDCl_3 ; (d) 3:1 mixture of mer- and $\text{fac-}[\text{Fe}(\text{fbpy})_3](\text{PF}_6)_2$ in $(\text{CD}_3)_2\text{CO}$. Chemical shifts were measured relative to an external sample of pyridine in $(\text{CD}_3)_2\text{CO}$ and converted to the nitromethane scale by addition of 61.79 p.p.m.⁵

a doublet ('fluoropyridyl') peak, with $^4J(\text{FN})$ 6 Hz. Figure 1(d) shows the spectrum of a 3:1 mixture of mer- and $\text{fac-}[\text{Fe}(\text{fbpy})_3]^{2+}$. From symmetry considerations, three singlets ('pyridyl') and three doublets ('fluoropyridyl') are expected for the six non-equivalent nitrogens in $\text{mer-}[\text{Fe}(\text{fbpy})_3]^{2+}$, and one singlet and one doublet for the two sets of non-equivalent nitrogen atoms in $\text{fac-}[\text{Fe}(\text{fbpy})_3]^{2+}$; Figure 1(d) confirms the presence of a total of four singlet and four doublet nitrogen peaks, of approximately equal intensity. These results demonstrate that each of the six non-equivalent nitrogen atoms in $\text{mer-}[\text{Fe}(\text{fbpy})_3]^{2+}$ can be observed by ^{15}N n.m.r. spectroscopy. In view of the continuing interest in transition metal-bipyridyl systems,⁴ the fbpy ligand, combined with ^{15}N and ^{19}F n.m.r. spectroscopy, may be useful in further stereochemical and mechanistic studies of these systems.

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